Vol. 07, Issue, 04, pp.8255-8259, April 2025 Available online at http://www.journalijisr.com SJIF Impact Factor 6.599

# **Research Article**



# FORMULATION AND EVALUATION OF FAST DISINTEGRATING TABLETS OF AMILORIDE HYDROCHLORIDE USING DIFFERENT SUPER DISINTERGENTS

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#### Received 15th February 2025; Accepted 16th March 2025; Published online 30th April 2025

#### ABSTRACT

Amiloride Hydrochloride (Amiloride HCL) is a potassium sparing diuretic and antihypertensive agent, used to treate hypertension. The objective of the present study was to formulate and evaluate fast dissolving tablets of Amiloride HCL Fast dissolving drug delivery system offers a solution for those patients having difficulty in swallowing, attempt was been made to prepare fast dissolving tablets of Amiloride hydrochlorideusing super disintegrants like sodium starch glycolate and croscarmellose sodium and the prepared granules were subjected D.C method. The granules thus prepared were evaluated for hardness, friability, weight variation, in vitro disintegration time and in vitro dissolution study. Compatibility of the drug and the excipients were studied by FTIR spectroscopy ,from the results obtained no interaction between the drug and excipients were observed. From this study it was concluded that fast dissolving tablets of Amiloride hydrochloride prepared from three different superdisintergents enhanced dissolution will lead to improved bioavailability, improved the effectiveness of Amiloride hydrochloride.

Keywords: Amiloride hydrochloride, super disintegrates and fast disintegrating tablets.

# **INTRODUCTION**

The fast-dissolving tablets are becoming popular as novel delivery systems for drug administration. They are more convenient for children, elderly patients, patients with swallowing difficulties, and in the absence of potable liquids. It is one of the most desirable formulation for the elderly as it is easy to swallow easy to handle. Taking these requirements into consideration, attempts have been made to develop a fast-disintegrating tablet. Since such a tablet can disintegrate in only a small amount of water in the oral cavity.

However, disintegrating agents are the substances routinely included in the tablet formulations to aid in the break-up of the compacted mass into the primary particles to facilitate the dissolution or release of the active ingredients when it is put into a fluid environment. They enhance moisture penetration and dispersion of the tablet matrix. The major function of disintegrants is to oppose the efficiency of the tablet binder and physical forces to act under compression to form the tablet. Recently new materials termed as "superdisintegrants" have been developed to improve the disintegration processes.

Moreover, superdisintegrants are another version of super-absorbing materials with swelling properties as required. These materials swell quickly and are more effective at lower concentrations with greater disintegrating efficiency and mechanical strength. Superdisintegrants are generally used at a low quantity in the solid dosage form, typically 1 - 10% by weight relative to the total weight of the dosage unit. Generally, one gram of superdisintegrant absorbs 10-40 g of water or aqueous medium. After absorption, swelling pressure and isotropic

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swelling of the superdisintegrant particles create stress concentrated areas where a gradient of mechanical properties will exist due to which whole structure will break apart.



Figure 1: Superdisintegrants breaking the tablet structure

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Figure 2: Manufacturing techniques for FDTs

# **MATERIALS AND METHODOLOGY**

Amiloride Hydrochloride was obtained from Sun pharma Mumbai croscarmellose sodium. sodium starch glycolate was gift sample from Astra Zeneca Bangalore and other chemical obtained from laboratory.

# FORMULATION OF AMILORIDE HCL FAST DISSOLVING TABLETS

#### Method for preparation of FDTs:

Tablets containing 10 mg of Amiloride Hydrochloride were prepared by direct compression technique. The drug and all other excipients, except sodium stearyl fumarate were previously sieved through a 60 mesh were accurately mixed for 15 min by using a poly bag. The resulting mixture was mixed with sodium stearyl fumarate and combined for mixing, up to 5 min. The mixer was compressed by using 10mm concave punches on sixteen station rotary tablet compression machine.

Table.1 General Compositi	ion of Formulation Prepa	red by Direct Com	pression Method
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Formula	Drug	SSG	CMS	Microcrystalline cellose	Mannitol	Aspartame	Sodium stearyl fumarate	Talc
F1	10mg	4mg	-	30mg	56mg	4mg	1mg	1 mg
F2	10mg	6mg	-	30mg	54mg	4mg	1mg	1 mg
F3	10mg	8mg	-	30mg	52mg	4mg	1mg	1mg
F4	10mg	10mg	-	30mg	50mg	4mg	1mg	1 mg
F5	10mg	12mg	-	30mg	48mg	4mg	1mg	1 mg
F6	10mg	-	4mg	30mg	56mg	4mg	1mg	1 mg
F7	10mg	-	6mg	30mg	54mg	4mg	1mg	1 mg
F8	10mg	-	8mg	30mg	52mg	4mg	1mg	1 mg
F9	10mg	-	10mg	30mg	50mg	4mg	1mg	1 mg
F10	10mg	-	12mg	30mg	48mg	4mg	1mg	1 mg
Formula	Drug	SSG	CMS	Microcrystalline cellose	Mannitol	Aspartame	Sodium stearyl fumarate	Talc
F1	10mg	4mg	-	30mg	56mg	4mg	1mg	1 mg
F2	10mg	6mg	-	30mg	54mg	4mg	1mg	1 mg
F3	10mg	8mg	-	30mg	52mg	4mg	1mg	1mg
F4	10mg	10mg	-	30mg	50mg	4mg	1mg	1 mg
F5	10mg	12mg	-	30mg	48mg	4mg	1mg	1 mg
F6	10mg	-	4mg	30mg	56mg	4mg	1mg	1 mg
F7	10mg	-	6mg	30mg	54mg	4mg	1mg	1 mg
F8	10mg	-	8mg	30mg	52mg	4mg	1mg	1 mg
F9	10mg	-	10mg	30mg	50mg	4mg	1mg	1 mg
F10	10mg	-	12mg	30mg	48mg	4mg	1mg	1 mg

### ESTIMATION OF AMILORIDE HYDROCHLORIDE

#### Development of calibration curve for Amiloride hydrochloride. Preparation of Standard Stock Solution

A standard stock solution of Amiloride HCL was prepared by dissolving accurately weighed 100mg of Amiloride HCL in phosphate buffer (pH 6.8) to obtain volume upto 100ml in volumetric flask (STOCK I).

#### Preparation of working standard solutions

From stock solution appropriate aliquote were pippette into different volumetric flask and volumes were made up to 10ml with phosphate buffer (pH 6.8) solution so as to get drug concentration of 2, 4, 6, 8 and  $10\mu$ g/ml. The absorbance of these drug solutions were estimated at 362nm.

#### Drug to polymer compatibility study's:

IR spectrum of pure drug and other excipients were recorded using FTIR spectrophotometry.

#### **Pre-compression parameters**

**Angle of Repose:** The angle of repose is the constant, threedimension angle (relative to horizontal base) assumed by a cone like pile of material formed by any of several different methods. When the angle of repose exceeds 50degrees, the flow is rarely accepted for manufacturing purpose.

**Bulk Density:** The bulk density was determined by transferring the accurately weighed sample of powder to the graduate cylinder. The initial volume and weight were noted. Ratio of weight of the sample was calculated by using the following formula. Bulk Density =Mass /Bulk volume.

**Tapped Density:** Weighed powder sample was transferring to a graduated cylinder and was placed on the tap density apparatus, was operated for fixed number of taps (100). The tapped density was determined by the following formula.

Tapped Density =Mass/Tapped volume.

**Percentage compressibility**: Based on the apparent bulk density and tapped density, the percentage compressibility of the bulk drug was determined by the following formula

%Compressibility = Tapped Density-Bulk density×100/Tapped density.

Hauser's Ratio: It is measured by the tapped density to bulk density. Hauser's Ratio= Tapped density/Bulk density

# **POST COMPRESSION PARAMETERS**

### Tablet thickness:

Randomly 5 tablets were taken from each formulation trial batch and their thickness was determined by using screw gauge.

#### Weight variation test:

20 tablets were randomly selected from each formulation trial batch and their average weight was calculated using digital balance. Individual weight of each tablet was also calculated using the same and compared with the average weight.

#### Measurement of tablet hardness:

Hardness of 10 tablets was found using Monsanto hardness tester, mean and standard deviation were computed and reported. It is expressed in kg/cm  $^{\rm 2}$ 

#### Friability:

10 tablets were weighed and placed in Roche friabilator where the tablets were exposed to rolling and repeated shocks resulting from free falls within the apparatus. The Friabilator was operated at 25 rpm for 4 mins. After 100 revolutions, tablets were removed, deducted and weighed again. The friability was determined as the percentage loss in weight of the tablets.

#### **Disintegration Time:**

The test was carried out on the 3 tablets using the apparatus specified in USP distilled water at 37  $^{\circ}$  C  $\pm$  2  $^{\circ}$  C was used as a disintegration media and the time in seconds taken for complete disintegration of the tablet with no palpable mass remaining in the apparatus was measured in seconds.

#### Wetting time:

A sample of the final tablet was placed in a petri dish (10 cm in diameter) containing 10 ml water at room temperature. The wetting time is the time necessary for the complete wetting of the tablet is noted.

#### In vitro dissolution:

Freshly prepared 0.1 NHCl of 900 ml was placed in each dissolution vessels of dissolution test apparatus (USP, II paddle method). The tablets were placed in the dissolution medium. The temperature of the dissolution medium was maintained at  $37 \pm 0.5^{\circ}$ C and the paddle was rotated at 50 rpm. 5 ml samples were withdrawn. The sample volume was immediately replaced with the same volume of fresh media as when a sample was taken. The samples withdrawn were filtered, diluted and estimated spectrophotometrically at 362nm.

#### Water absorption ratio

Apiece of tissue paper folded twice was placed in a small petri dish containing 6 ml of water. A tablet was put on the paper and time required for complete wetting was measured. The wetted tablet was then weighed. Water absorption ratio, R was determined using following equation.

R=100 (Wa -Wb)/Wb

Where,

Wb- weight of tablet before absorption Wa -weight of tablet after absorption Three tablets from each formulation were tested. Average and standard deviation were determined.

### Uniformity of drug content:

Ten tablets from each batch were weighed accurately and powdered. An amount of powder equivalent to 4mg of the drug was transferred into a 25ml volumetric flask. The volume was made with 6.8 pH phosphate buffer and sonicated for 10 min. The resulting solution was filtered and assayed at 362nm using spectrophotometer and drug content per tablet was determined.

# **RESULT AND DISSCUSSION**



#### Fig No 3 FTIR Spectra of Pure Drug



Fig No 4: FTIR spectra of Pure Drug and Excipients.

Formulations	Angle of repose I	Bulk Density (g/cc)	Tapped Density (g/cc)	% Carr's index	Hausner's Ratio %
F1	21.5± 0.270	0.481±1.295	0.509±0.915	4.98±1.759	1.05±0.656
F2	22.61±0.813	0.484±0.48	0.512±0.974	5.46±1.245	1.057±0.128
F3	20.91±0.270	0.490±1.295	0.509±0.915	3.16±1.759	1.03±0.956
F4	21.51±0.270	0.479±1.295	0.506±0.915	5.4±1.759	1.057±0.056
F5	23.22±0.257	0.568±0.421	0.625±0.730	9.12±1.350	1.10±0.164
F6	23.21±0.270	0.501±1.295	0.60±0.915	16.6±1.759	1.2±0.556
F7	22.22±0.270	0.512±1.275	0.610±0.912	16.68±1.759	1.17±0.520
F8	23.14±0.270	0.449±1.230	0.540±0.315	7.58±1.650	1.19±0.530
F9	21.23±0.270	0.512±1.394	0.622±0.615	17.68±1.732	1.08±0.556
F10	23.23±0.270	0.533±1.394	0.628±0.615	17.75±1.732	1.77±0.556

#### Table No. 2 Data for Pre-Compression Parameters.

#### Table No.3: Data for Post -Compression Parameters

Formulation	Weight Variation(mg)	Hardness (Kg/cm²)	Friability (%)	Thickness (mm)	Uniformity of Content (%)	Water absorption ratio(%)	Disintegration Time (seconds)
F1	106±1.35	2.8±0.98	0.24±0.89	2.95±0.23	90.78±0.14	89.91±0.96	44.50±0.13
F2	106±0.97	2.7±0.29	0.36±0.04	2.28±0.12	91.48±0.13	93.48±0.59	41.2±0.11
F3	106±0.43	2.8±0.16	0.32±1.2	2.32±0.54	93.62±0.23	94.58±0.45	39.23±0.23
F4	106±1.56	2.9±1.16	0.25±0.69	2.41±0.43	92.23±0.06	91.98±0.49	32.31±0.03
F5	106±1.25	2.8±0.26	0.36±0.48	2.34±0.13	93.57±0.01	97.32±0.48	31.23±0.12
F6	106±1.09	2.9±0.64	0.34±0.18	2.31±0.25	95.83±0.02	98.25±0.75	35.24±0.14
F7	106±1.22	2.9±0.9	0.22±0.8	2.28±0.23	97.52±0.05	97.62±0.98	20.21±0.23
F8	106±1.82	2.7±0.8	0.34±0.4	2.29±0.18	98.15±0.18	98.48±0.87	19.02±0.43
F9	106±2.4	2.8±0.9	0.37±0.6	2.76±0.16	99.42±0.24	99.58±0.84	18.22±0.19
F10	105±2.4	2.76±0.9	0.38±0.6	2.88±0.16	97.42±0.24	97.58±0.84	20.22±0.19

	Table n	o 4: Data	for Disso	olution	Profile
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Time (min)	Cumulative %Drug released									
	F1	F2	F3	F4	F5	F6	F7	F8	F9	F10
0	0	0	0	0	0	0	0	0	0	0
5	9.2	9.42	9	9.55	9.23	9.42	9.81	9.8	9.4	9.33
10	27.8	25.81	28.45	27.28	20.26	26.51	20.25	24.6	22.4	20.4
15	55.6	49.57	50.25	46.58	38.49	44.72	35.33	44.6	44.8	38.8
20	60.75	65.95	67.77	65.44	56.45	63.88	50.32	61.75	64.2	63.2
25	82.68	74.83	78.34	76.73	65.37	76.83	65.66	74.28	76.4	77.4
30	92.56	82.74	88.92	86.36	77.69	87.98	70.52	82.92	85.2	86.2
45	93.93	91.41	92.35	92.97	85.69	90.36	95.32	94.92	88.6	87.6
60	94.67	92.98	93.02	93.26	91.96	91.42	96.38	97.92	99.2	93.2

#### RECOMMENDATIONS

This study can be done on large scale by varying the choice of Super Disintegrants and Evaluate the performance of various super disintegrants such as Sodium Starch Glycolate (SSG), Croscarmellose Sodium (CCS), Polyvinylpyrrolidone (PVP) and Microcrystalline Cellulose (MCC)

It will test different concentrations (e.g., 2%, 4%, 6%, 8%) of each super disintegrant to determine the optimal level that provides the fastest disintegration and dissolution without compromising mechanical strength.

#### CONCULSIONS

In the present study, Amiloride hydrochloride was used as a model drug. Fast dissolving tablets containing Amiloride hydrochloride were prepared by direct compression technique. Superdisintegrants used were added each in different concentrations to formulate 10 formulations. The prepared tablets of Amiloride hydrochloride were evaluated for pre-compression parameters like angle of repose, bulk density, tapped density, Carr's index, Hausner's ratio and post-compression parameters like hardness, thickness, friability, weight variation, wetting time, water absorption ratio, drug content, *in-vitro* disintegration time and *in-vitro* dissolution studies.

The observations have shown that all the FDT formulations were acceptable with reasonable limits of standards required for fast dissolving tablets. The study reveals that the formulations prepared by Croscarmellose was best than Sodium starch glycolate .Formulation prepared by Croscarmellose F9 was the best formulation. The study revealed that superdisintegrants used were effective in low concentration level.

Thus, it was concluded that the fast dissolving tablets containing Amiloride hydrochloride was successfully prepared by direct compression technique and characterized.

#### Acknowledgments

I would like to express my grateful thanks to our chairman Dr. M.R.Huinayakar for his guidance and support. My deepest gratitude to our director Dr Gopal krishna Shridevi Institute of Pharmaceutical Sciences Tumakuru for Valuable suggestions & guidance. Thanks to Dr.TS. Bheemaraju HOD. of Psychiatric Nursing for ho valuable suggestions & guidance for manuscript publication.

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